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Key indicators

Single-crystal X-ray study

$T = 180\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

R factor = 0.037

wR factor = 0.096

Data-to-parameter ratio = 17.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Azelaic acid

The crystal structure of heptane-1,7-dicarboxylic acid (azelaic acid), $\text{C}_9\text{H}_{16}\text{O}_4$, has been redetermined at 180 K. The molecular units have twofold symmetry and are linked *via* the ubiquitous *syn-syn* carboxylic acid dimer to form infinite chains running approximately along the $[\bar{4}01]$ vector.

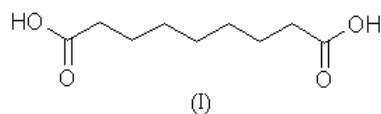
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Comment

Two polymorphs of azelaic acid, (I), have been reported previously: the α form crystallizes in $P2_1/c$ (Caspari, 1928; Housty & Hospital, 1967) and the β form crystallizes in $C2/c$ (Housty & Hospital, 1967). For both polymorphs, the structures present in the CSD (AZELAC10 and AZELAC01; Allen & Kennard, 1993) are derived from room-temperature



data with R factors *ca* 10% and ambiguities in the treatment of H atoms. We have, therefore, re-examined azelaic acid and report here the structure of the β form measured at 180 K to significantly greater precision.

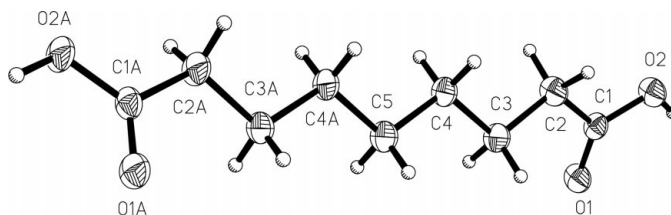


Figure 1

The molecular unit of the title compound showing displacement ellipsoids at the 50% probability level.

Experimental

Azelaic acid was obtained from Aldrich and recrystallized from ethanol.

Crystal data

$\text{C}_9\text{H}_{16}\text{O}_4$

$M_r = 188.22$

Monoclinic, $C2/c$

$a = 22.622(2)\text{ \AA}$

$b = 4.7348(2)\text{ \AA}$

$c = 9.6864(7)\text{ \AA}$

$\beta = 110.559(3)^\circ$

$V = 971.5(1)\text{ \AA}^3$

$Z = 4$

$D_x = 1.287\text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Cell parameters from 2161

reflections

$\theta = 1.0\text{--}27.5^\circ$

$\mu = 0.10\text{ mm}^{-1}$

$T = 180(2)\text{ K}$

Plate, colourless

$0.25 \times 0.12 \times 0.06\text{ mm}$

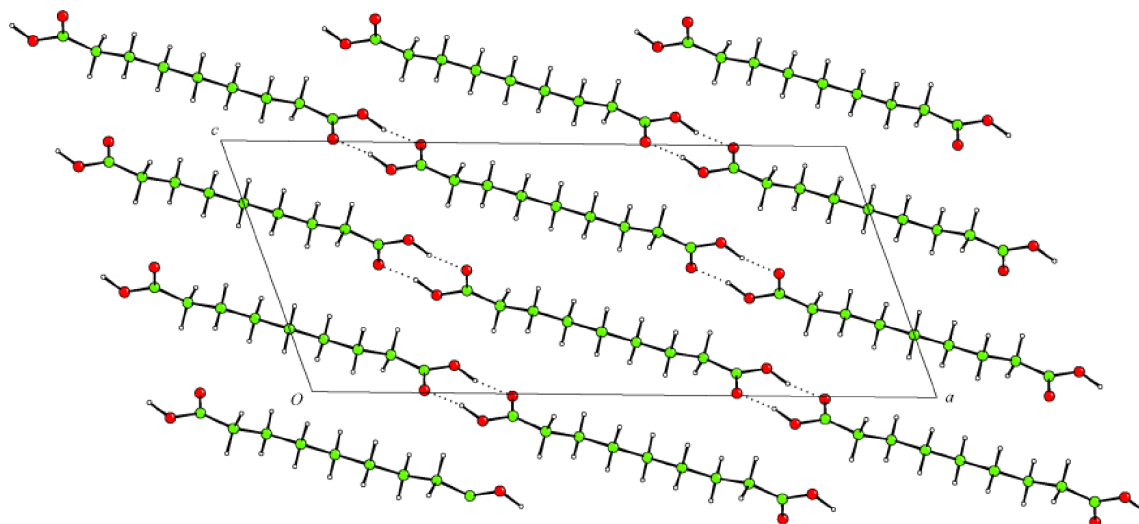


Figure 2
Projection onto (010) showing hydrogen-bonded chains running approximately along the $[\bar{4}01]$ vector.

Data collection

Nonius KappaCCD diffractometer
Thin-slice ω and φ scans
Absorption correction: multi-scan
(*SORTAV*; Blessing, 1995)
 $T_{\min} = 0.907$, $T_{\max} = 0.994$
3154 measured reflections
1085 independent reflections

913 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 27.4^\circ$
 $h = 0 \rightarrow 28$
 $k = -6 \rightarrow 6$
 $l = -12 \rightarrow 11$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.096$
 $S = 1.09$
1085 reflections
64 parameters
H atoms: see below

$w = 1/[\sigma^2(F_o^2) + (0.0328P)^2 + 0.5150P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.010$
 $\Delta\rho_{\text{max}} = 0.25 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.19 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O2-H1 \cdots O1^i$	0.96 (2)	1.70 (2)	2.6576 (12)	173.5 (17)

Symmetry code: (i) $\frac{1}{2} - x, \frac{1}{2} - y, -z$.

The H atom of the carboxylic acid group was located in a difference Fourier map and refined without restraint. All other H atoms were placed geometrically and allowed to ride during subsequent

refinement with an isotropic displacement parameter fixed at 1.2 times that for the C atom to which they are attached.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Sheldrick, 1993) and *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *SHELXL97*.

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